

RAMAKRISHNA MISSION VIDYAMANDIRA

(Residential Autonomous College under University of Calcutta)

B.A./B.Sc. FIFTH SEMESTER EXAMINATION, DECEMBER 2014

THIRD YEAR

INDUSTRIAL CHEMISTRY (Honours)

Date : 20/12/2014

Time : 11 am – 1 pm

Paper : V (Gr. B)

Full Marks : 50

Unit – I

1. Write notes on **any four** :

[4×5]

- Bhopal gas accident
- BOD, COD & TOC
- Green house effect
- E1A
- Hydrosphere / Water cycle
- Automobile pollution

Unit – II

(Answer **any five** questions)

[5×6]

- Draw the potential energy curve for n-butane molecule and discuss about the relative stability of the conformations? [3]
 - Write the most stable conformation for 1,2-di hydroxyl ethanol and explain the reason for stability. [1.5]
 - What is torsion angle and how it differs from dihedral angle? [1.5]
- What do you mean by Cs pathway of inversion of cyclohexane chair form? Explain with diagram. [3]
 - Explain 1,3 diaxial interaction for 1,3 dimethyl cyclohexane and draw the most stable conformation. [3]
- Which conformer of 1,2-dicarboxyl cyclohexane can form acid anhydride easily? Explain. [2]
 - Trans-4-tert-butylcyclohexanol undergoes chromic acid oxidation quickly than its cis-isomer —explain [2]
 - When 4-tertiarybutyl cyclohexanone is treated with LiAlH_4 , explain the formation of major product. [2]
- Discuss briefly about Bardhan-Sengupta synthesis of phenanthrene and give mechanism and reagent used in each step. [6]
- Explain the mechanistic view of electrophilic aromatic nitration during the nitration reaction of naphthalene. Also explain the formation of major isomer in this reaction. [6]
- Discuss about Hoffmann's method of separation of amines. [2]
 - Compare the basicity of $\text{CH}_3\text{CH}_2\text{NH}_2$, $(\text{CH}_3\text{CH}_2)_2\text{NH}$ and $(\text{CH}_3\text{CH}_2)_3\text{N}$ in aqueous as well as non-aqueous medium with proper explanation. [4]
- Do the conversion (**any three**) : [3×2]
 - Nitrobenzene to *m*-nitro aniline
 - 1,4-naphthaquinone to anthracene
 - Aniline to di-phenyl hydrazine
 - Nitrobenzene to *p*-amino phenol
 - N-methylphthalimide to methyl amine